Lecture notes

Wednesday, 21 March 2018

Tasks for next week:

1:05 PM

- Read chapter 4 (Kittel): Phonons/ Lattice Vibrations

Crystal bindings

Alon 1
- There is repulsive interaction Vrepulsive
- Altractive interaction Valeractive V= Vrepulsive + Valtractive

-> Crystel: Many body problem: Viole & Z Vi

(x) = Vrepulsive + Vattactive

Ro "most favourable point for distance between atoms" -> find $\frac{dV(x)}{dx} = 0 => x = R_0$ V(Ro) "binding potubial" -> binding energy

Attractive Interactions: -van der Waals

- louic birding - covalent binding - metallia

- What about hydrogen bounds? Nobel gases at very low temperature form crystals! Binding force in nobel gas crystal: van des Waals

L) form fee crystals Van des Wagls interactions Two dipoles

oscillators with spring constant C $\downarrow \qquad \qquad \geqslant$ Hamiltonian: Ho= + 1/2 Cx2 + P2 + 1/2 Cx2 "un compled" La Eigenstates of a harmonic oscillator: E= two (n+2)

- both considered harmonic

=> E = うちい

Hamiltonian for the coupled system: $H_1 = \frac{e^2}{R} + \frac{e^2}{R + x_1 - x_2} - \frac{e^2}{R + x_1} - \frac{e^2}{R - x_2}$

Assumption: R>> x1, x2 "Distance between atoms much bigger than length of dipole"

La Taylor expansion: $\frac{1}{1-x} = 1+x+x^2+\sigma(x^3)$ $H_1 = \frac{e^2}{R} \left[1 + \left(\frac{1}{1 - \frac{x_1 - x_1}{R}} \right) - \left(\frac{1}{1 - \frac{x_2}{R}} \right) - \frac{1}{1 - \frac{x_2}{R}} \right]$

 $=\frac{e^2}{R}\left[1+\left(1+\frac{x_2-x_1}{R}\right)+\left(\frac{x_2-x_1}{R}\right)^2\right)-\left(1+\left(\frac{x_1}{R}\right)+\left(-\frac{x_1}{R}\right)^2\right)-\left(1+\left(\frac{x_2}{R}\right)+\left(\frac{x_2}{R}\right)^2\right)$

 $=\frac{e^2}{R}\left[-\frac{2\times_1\times_2}{R^2}\right]=-\frac{2e^2\times_1\times_2}{R^3}$ Hamiltonian H= Ho+H, = P2 - + 1 Cx2 + P22 - 2e2x1x2

Now use toich: $x_s = \frac{x_1 + x_2}{\sqrt{x_1}}$ $x_a = \frac{x_1 - x_2}{\sqrt{x_1}}$

b) $x_1 = \frac{x_3 + x_\alpha}{\sqrt{27}}$ $x_2 = \frac{x_3 - x_\alpha}{\sqrt{27}}$ —s to add H_o and H₁ together For the monestum: $P_1 = \frac{P_S + P_A}{\sqrt{27}}$ $P_2 = \frac{P_S - P_A}{\sqrt{27}}$

 $=) H = \frac{(\rho_{s} + \rho_{a})^{2}}{4 \mu_{s}} + \frac{(\rho_{s} - \rho_{a})^{2}}{4 \mu_{s}} + \frac{(x_{s} + x_{a})^{2}}{4} + \frac{(x_{s} + x_{a})^{2}}{4} + \frac{(x_{s} + x_{a})^{2}}{4} - \frac{e^{2}(x_{s}^{2} - x_{a}^{2})}{6}$

 $= \frac{R^{2} + R^{2}}{2m} + \frac{C(x_{s}^{2} + x_{a}^{2})}{2} - \frac{e^{2}(x_{s}^{2} - x_{a}^{2})}{R^{3}} = \frac{R^{2}}{2m} + x_{s}^{2} \left(\frac{c}{2} - \frac{e^{2}}{R^{3}}\right) + \frac{R^{2}}{2m} + x_{a}^{2} \left(\frac{c}{2} + \frac{e^{2}}{R^{3}}\right)$ -) This is again the hamiltonian of a harmonic oscillator

Frequency: $\omega_{\pm}^{2} \left[\left(C \pm \frac{2e^{2}}{R^{3}} \right) / m \right]^{1/2} = \omega_{0} \left[1 \pm \frac{1}{2} \left(\frac{2e^{2}}{CR^{3}} \right) - \frac{1}{8} \left(\frac{2e^{2}}{CR^{3}} \right)^{2} + \dots \right]$ = 1/m Taylor expansion En = 2 = two = two

 $E = \frac{1}{2} t_1 \left(\omega_1 + \omega_2 \right) = \frac{1}{2} t_1 \left(2 \omega_0 - \frac{\omega_0}{4} \left(\frac{2e^2}{CR^3} \right)^2 \right)$

Difference between the energy of the uncoupled and coupled dipol system - this is the binding energy $\Delta U = E - E_0 = \hbar \omega_0 - \frac{\hbar e^4 \omega_0}{2 c^2} \cdot \frac{1}{R^6} - \hbar \omega_0 = -\frac{\hbar e^4 \omega_0}{2 c^2} \frac{7}{R^6}$ DU Q -R-6

Lennard - Jones Potential $U(R) = \frac{B}{R^{12}} - \frac{A}{R^6} = 4 \varepsilon \left[\left(\frac{\sigma}{R} \right)^R - \left(\frac{\sigma}{R} \right)^6 \right]$

Where B= 48012 and A= 4806 Full crystal

 $U_{\text{Total}} = 4 \varepsilon \frac{1}{Z} \sum_{i} \left(\left(\frac{\sigma}{\rho_{ij} R} \right)^{1/2} - \left(\frac{\sigma}{\rho_{ij} R} \right)^{6} \right)$ Assumption: crystel so big that you can take only one

atom and account for all its interactions N = # atoms

PijR = distance between atom i and j

R, R2, R3, ... = nearest, next-nearest, next-next-nearest, ... neighbour distance $K_i = \frac{R_i}{R}$

N; = # number of neighbours of distance R;

with this $\sum_{j}^{n} (P_{ij}R)^{-n} = \sum_{j}^{n} N_{ij} \cdot R_{j}^{n} = R_{j}^{n} \sum_{j}^{n} N_{ij} \times R_{j}^{n}$ They converge very fast!

Rewrite Utolal: Utolal = 2NE [An (5)2 + A6 (5)67 $\frac{d V_{\text{rotec}}}{d R} = 0 \implies \frac{R_{1} \log r}{\sigma} = \left(\frac{2 A_{12}}{A_{5}}\right)^{\frac{1}{6}}$

-> This model gives the information of why nobel gases condense to for