

PHY 127 FS2026

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Lecture 9

May 2026

Where were we?

• last lecture!

- quantum numbers of atoms
- electron transitions \rightarrow photons
- x-rays
- charged-particle detector
- Compton effect
- how photons lose energy

Penetration of x-rays

combined effect of Thomson scattering, photoelectric effect + Compton scattering generate attenuation of the X-ray beam.

$$I(x) = I_0 e^{-\mu \cdot x}$$

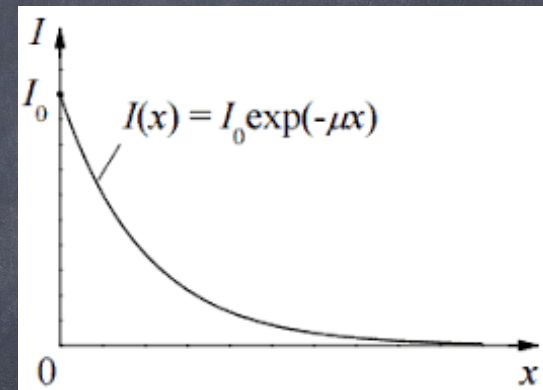
I_0 : initial beam intensity

$I(x)$: intensity at a depth, x

μ : attenuation coefficient with units $[m^{-1}]$

The mass attenuation coefficient, μ/ρ , is the attenuation per unit density of the material being penetrated.

Units: $[\frac{cm^2}{g}]$



photon energy	μ/ρ_{water}	$\mu/\rho_{dry\ air}$	μ/ρ_{bone}	μ/ρ_{muscle}	$\mu/\rho_{breast\ tissue}$
100 keV	0.17	0.15	0.18	0.16	0.16
10 keV	5.3	5.1	28	5.3	4.3
5 keV	43	40	190	42	34

Observations: The higher the x-ray energy, the farther the x-rays penetrate.
 (If M is large, attenuation is more, & the distance traveled is less)

W.O. • Chemical Bonds and X-Rays: Atomic and Molecular Physics 697

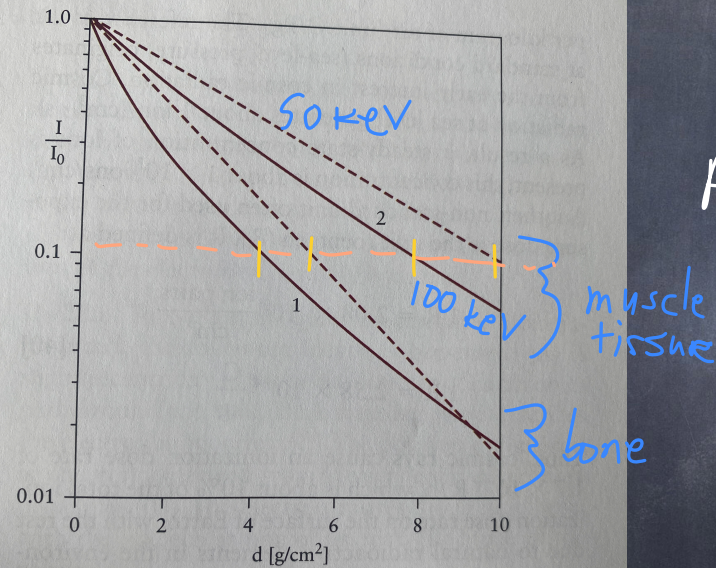


FIGURE 22.25

Intensity attenuation of X-ray beams of 100 keV (solid lines) and 50 keV (dashed lines) in human bones (curves 1) and muscle tissues (curves 2). The depth is given in unit g/cm^2 , which results from the product of density and the path length in the tissue.

The x-axis is given as the product of density ρ and the path length, x . $d = \rho x$

(Because this combination is more meaningful)
 For instance, x-ray intensity is reduced to 10% ($I/I_0 = 0.1$) ...

For muscle tissue, at $d \approx 8-10 \text{ g/cm}^2$

$$x = \frac{d}{\rho_{\text{muscle}}} \approx \frac{9 \text{ g/cm}^2}{1 \text{ g/cm}^3} = 9 \text{ cm}$$

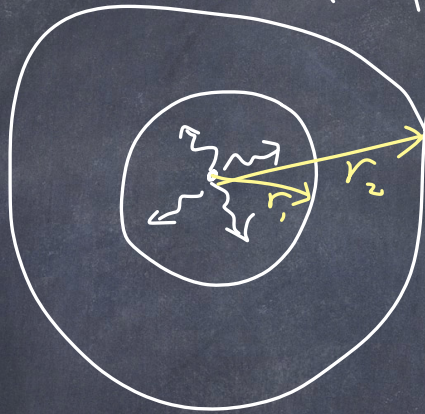
For bone, $d \approx 4.5-5.5 \text{ g/cm}^2$

$$x = \frac{d}{\rho_{\text{bone}}} \approx \frac{5 \text{ g/cm}^2}{1.2 \text{ g/cm}^3} \approx 4 \text{ cm}$$

From [1]: "Physics of the Life Sciences" by Martin Zinke-Allmang

Reminder:

$$\text{Intensity} = \frac{\text{Power}}{\text{area}}$$



Intensity will decrease like $\frac{1}{r^2}$
surface
Area of a sphere is $4\pi r^2$

If you have $r_2 = 2r_1$,
the intensity is 4 times less
at r_2 than r_1 ,

Where does the x-ray intensity go?

x-rays are either scattered or absorbed by bone or tissue.

Absorbed radiation has an adverse biological effect.

Measured radiation is reported in 2 ways:

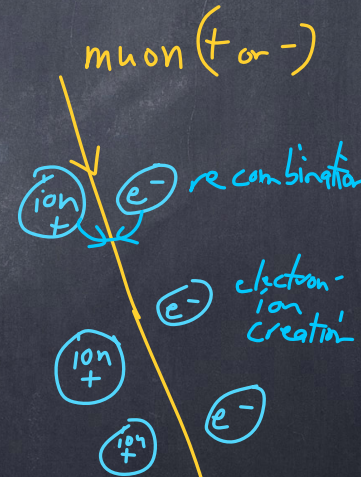
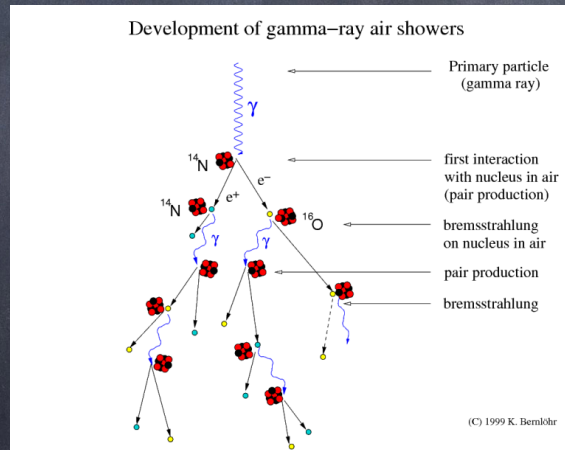
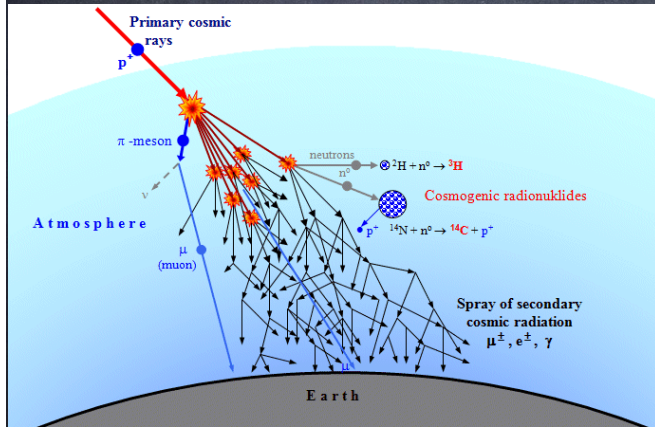
- 1) amount of ionization occurring in the material due to the radiation → exposure dose
- 2) energy deposited by radiation in the material → absorbed dose

Dose = total amount of ionization or energy deposited in a given amount of material.

Dose rate = dose per unit time.

There are different measures:
 exposure dose: total charge generated by
 ionization per kg of air
 units: $\left[\frac{C}{kg} \right]$ std. atmosphere

Cosmic radiation at sea level generates
 $1-10 \frac{\text{ions}}{\text{cm}^3/\text{s}}$. This equilibrium value is maintained
 despite recombination of ions.



Convert:

$$\frac{5 \text{ ions}}{\text{cm}^3/\text{s}} \cdot \left(\frac{100 \text{ cm}}{1 \text{ m}} \right)^3 \cdot \frac{1.6 \times 10^{-19} \text{ C}}{1 \text{ ion}} \cdot \frac{1 \text{ m}^3}{1.225 \text{ kg (density of air)}} = 6.54 \times 10^{-13} \frac{\text{C}}{\text{kg} \cdot \text{s}}$$

Another unit is roentgen (R)

$$1 \text{ R} = 2.08 \times 10^9 \frac{\text{ion pairs}}{\text{cm}^2}$$
$$= 2.58 \times 10^{-4} \frac{\text{C}}{\text{kg}}$$

$$= 6.54 \times 10^{-13} \frac{\text{C}}{\text{kg} \cdot \text{s}} \cdot \frac{1 \text{ R}}{2.58 \times 10^{-4} \frac{\text{C}}{\text{kg}}} \cdot \frac{3600 \text{ s}}{1 \text{ h}} \sim 9 \times 10^{-6} \frac{\text{R}}{\text{h}}$$

So cosmic rays cause ionization dose rate of $9 \times 10^{-6} \text{ R/h}$. This is about 10% of dose rate at earth's surface. The other 90% is natural radioactivity (Radon, ... ^{41}K (bananas)...))

More commonly used is the energy dose,
the energy deposited per kg of air
in units of $\left[\frac{\text{J}}{\text{kg}}\right]$

Units are called "gray"

$$1 \text{ gray} = 1 \text{ Gy} = 1 \frac{\text{J}}{\text{kg}}$$

Sometimes an older unit is the "rad"

$$1 \text{ Gy} = 100 \text{ rad}$$

$$1 \text{ R} \cong 1 \text{ rad} = 0.01 \text{ Gy}$$

Biological effect of radiation defined as the equivalent dose, $D_{\text{equivalent}}$, with units of sievert (Sv) defined so that the same value of D_{equiv} has the same impact on living tissue, for any type of radiation.

$$D_{\text{equiv}} = W_R \cdot W_T \cdot D_{\text{absorbed}}$$

radiation factor, expresses the physiological damage relative to t-ray radiation:

$W_R = 1$: for x-rays, electrons, positrons

$W_R = 5-10$: for neutrons

$W_R = 10$: alpha particles (He nucleus)

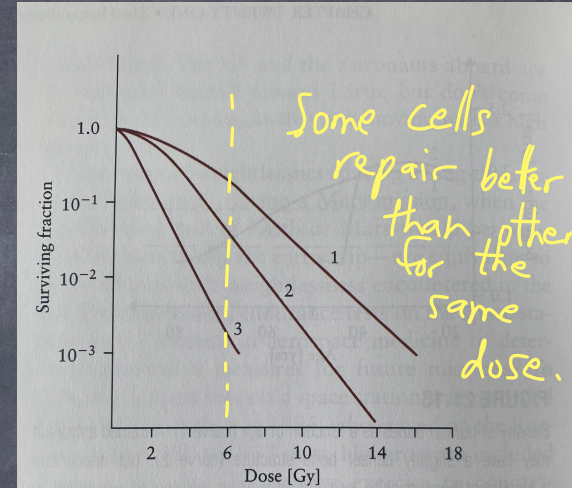


FIGURE 21.15

The surviving fraction of three types of human cells as a function of energy dose in unit Gy. The energy dose is the energy deposited by the radiation per kilogram of tissue. Note the lower steepness at doses below 1 Gy, which is due to self-repair mechanisms in living cells. Various cells respond with different sensitivity to radiation: (1) thyroid cells, (2) mammary cells, and (3) bone marrow.

[1] W_T : tissue weighting factor, would sum to 1 for the whole body, W_T is the damage compared to a whole body exposure
 D_{absorbed} : radiation exposure in Gray

Effect of dose

Equivalent dose
(Sv)

1-5

4-5

10-50

50-100

pathological diagnosis

serious temporary alterations of blood count

50% death rate in 30 days

Vomiting + nausea (die sooner)

brain + nerve damage (death in ~1 week)

acute dose
(all at once)

W_T : Tissue weighting Factor

	Female	Male
Testes	0	0.08
Ovaries	0.08	0
Bone surface	0.01	0.01
Bladder	0.04	0.04
Bone marrow, red	0.12	0.12
Brain	0.01	0.01
Breast	0.12	0.12
Colon	0.12	0.12
Liver	0.04	0.04
Lungs	0.12	0.12
Oesophagus	0.04	0.04
Salivary glands	0.01	0.01
Skin	0.01	0.01
Stomach	0.12	0.12
Thyroid	0.04	0.04
Remainder ^a	0.12	0.12

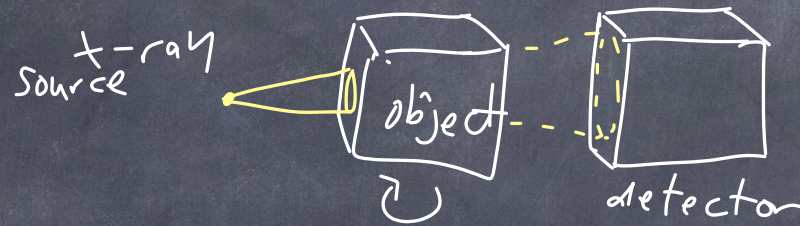
^aComponent organs for remainder in ICRP 103: adrenals, extrathoracic airways, gallbladder, heart, kidneys, lymphatic nodes, muscle, oral mucosa, pancreas, prostate, small intestine, spleen, thymus and uterus/cervix.

sum should add up to 1.

x-ray tricks for medical use

- 1) gastrointestinal tract can be imaged by x-rays if filled with dense material, Barium ($\rho = 3.5 \text{ g/cm}^3$) solution for increased contrast.
- 2) Similarly, iodine ($\rho = 4.93 \frac{\text{g}}{\text{cm}^3}$) + water, makes a soluble organic compound, used for cardiovascular system, urinary tract, + the brain.
- 3) mammography (lower energy x-rays, softer)
- 4) Improved images with computed tomography (CT) to obtain 3-D images from a collection of 2-D images, (x-ray images)

Originally, CT scan would have a single
x-ray source, and a detector opposite.

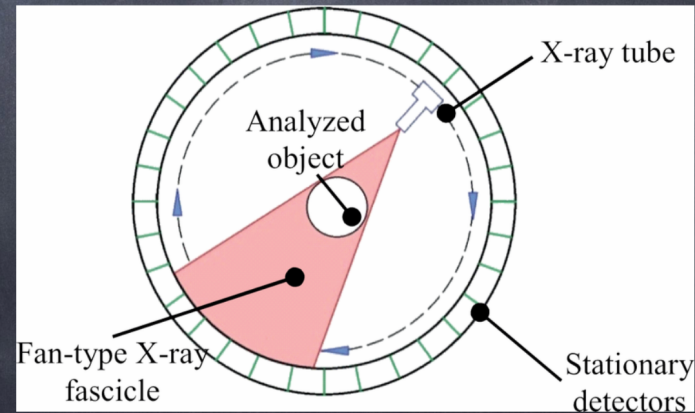
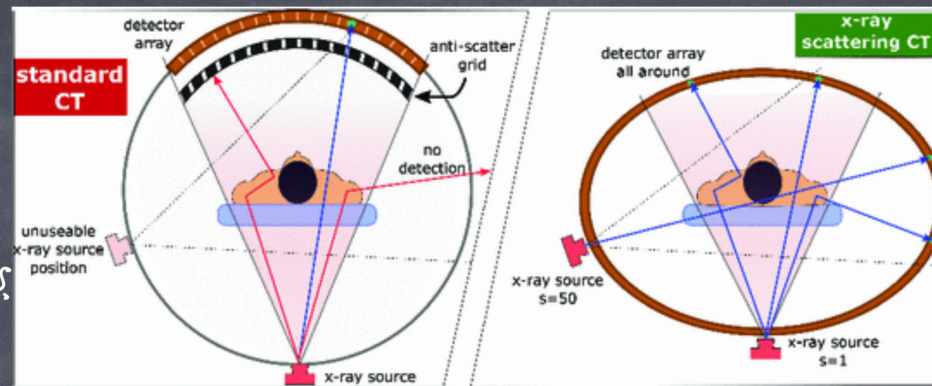


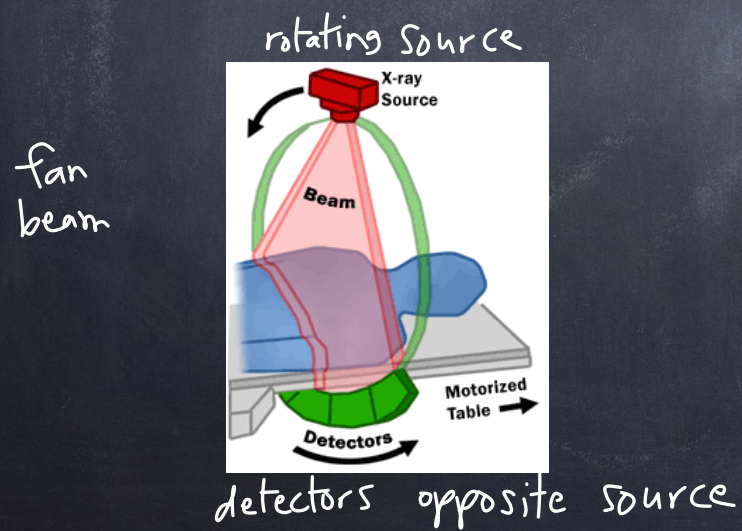
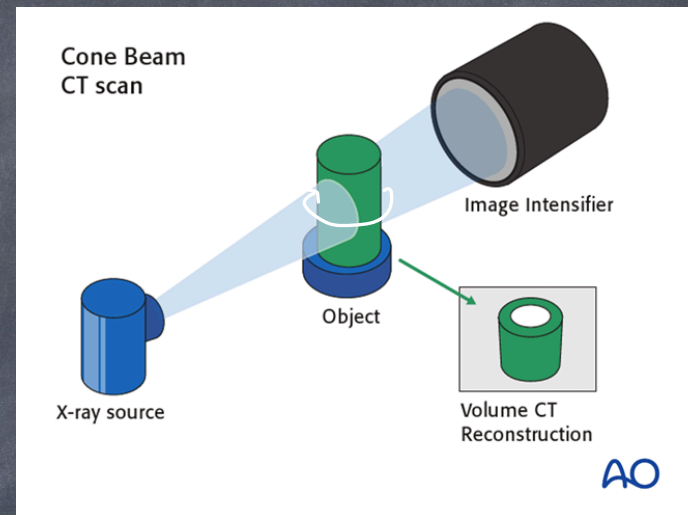
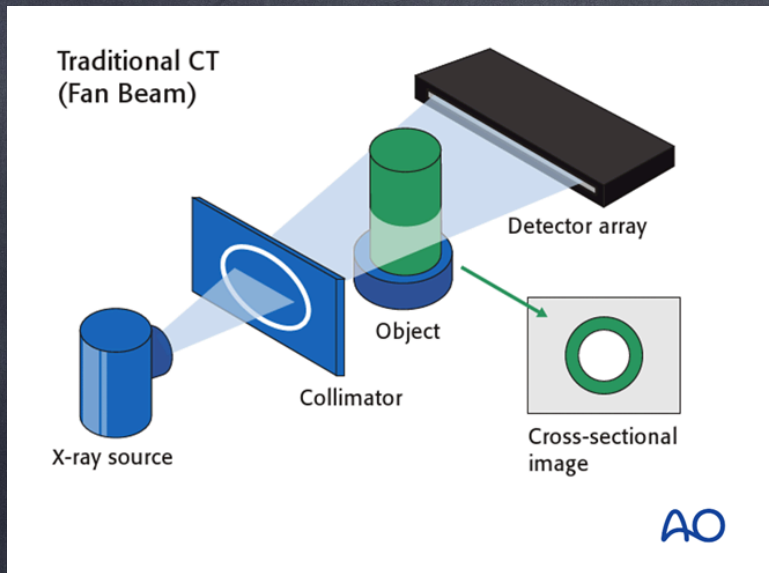
Rotate by 1° , take another x-ray,
 \sim minutes to do x-rays.



Today, wide fan-like beam, hundreds or thousands of detectors, \Rightarrow decreases time down to a few seconds.

Newest: stationary detectors, and the beam sweeps around the patient. Typically 50 ms per angle.





Black & white scale can be converted into color to represent brightness level, set according to absorption coefficient of tissue, μ , compared to water, μ_w .

Hounsfield unit

$$HU = 1000 \frac{\mu - \mu_{\text{water}}}{\mu_{\text{water}} - \mu_{\text{air}}}$$

OR

$$\text{CT number} = 1000 \frac{\mu - \mu_{\text{water}}}{\mu_{\text{water}}}$$

Since ρ_{air} is 800 times smaller than ρ_{water} CT number \sim HU

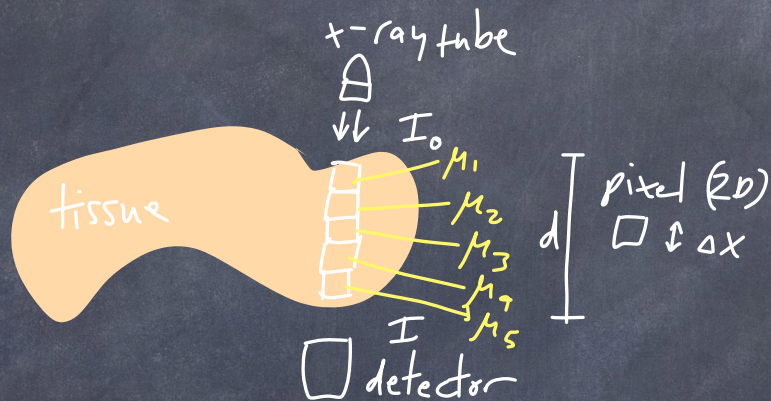
<u>material</u>	<u>CT number for x-rays of 60 keV</u>
water	0
air	-1000
bone	800
muscle	-48
Fat	-142



How to do and reconstruct a CT scan

x-ray beam goes through patient, different types of tissue are encountered, with different μ .

voxel (3D)



we have 5 unknowns:

$\mu_1, \mu_2, \mu_3, \mu_4, \mu_5$

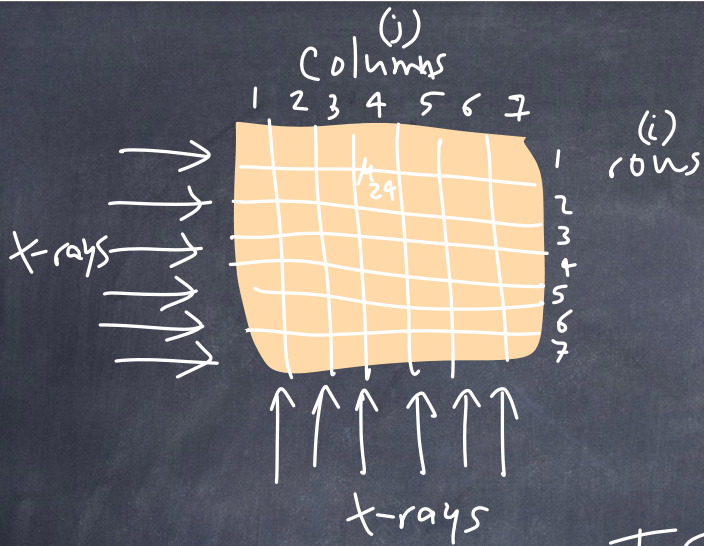
$$I = I_0 e^{-\sum_{i=1}^5 \mu_i \Delta x}$$

\uparrow we measure \uparrow we know

$$\log \frac{I_0}{I} = \sum_{i=1}^5 \mu_i \Delta x$$

If we make Δx smaller ($\Delta x \rightarrow 0$), then

$$I = I_0 e^{-\int_0^d \mu(x) dx}$$



We can measure I along different rows & columns

For the i th row

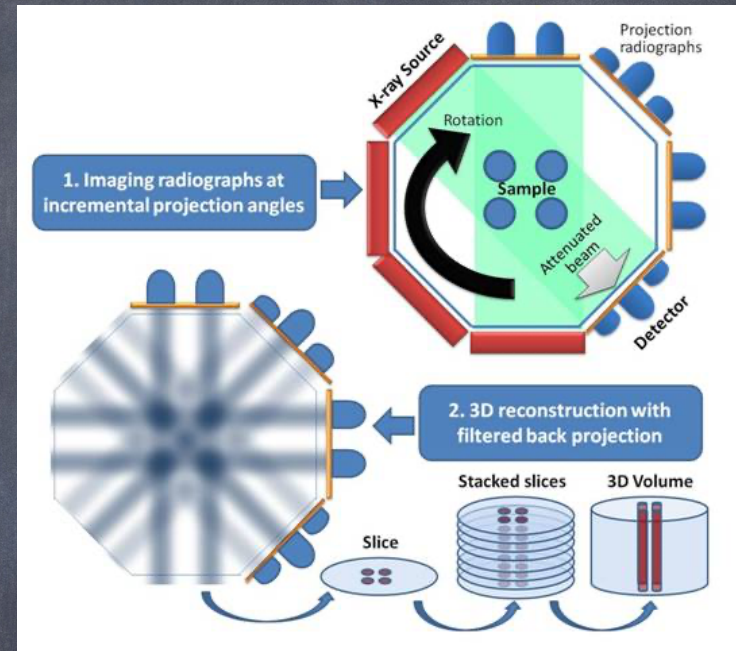
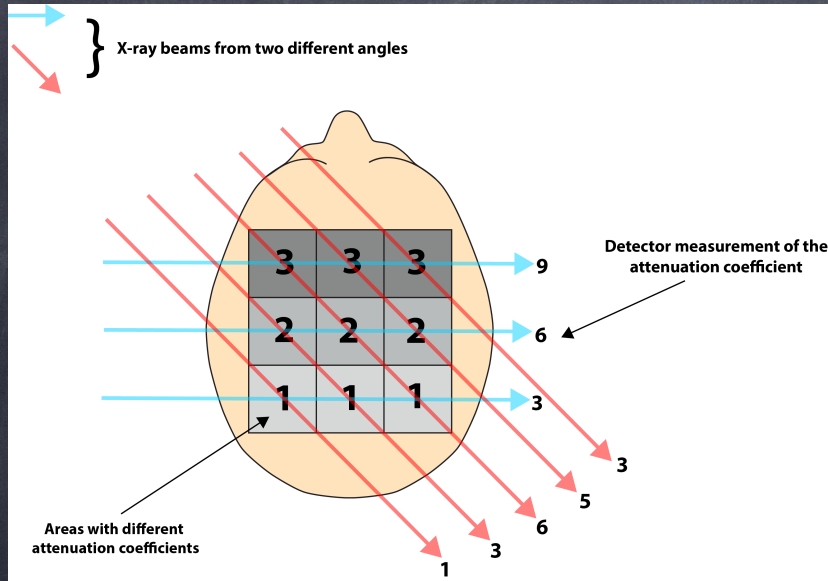
$$M_i(\Delta x) = \sum_{j=1}^7 M_{ij} \Delta x$$

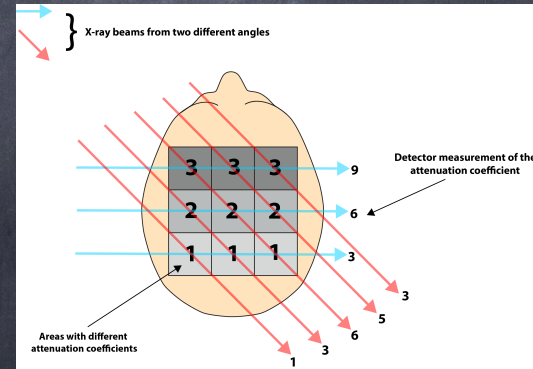
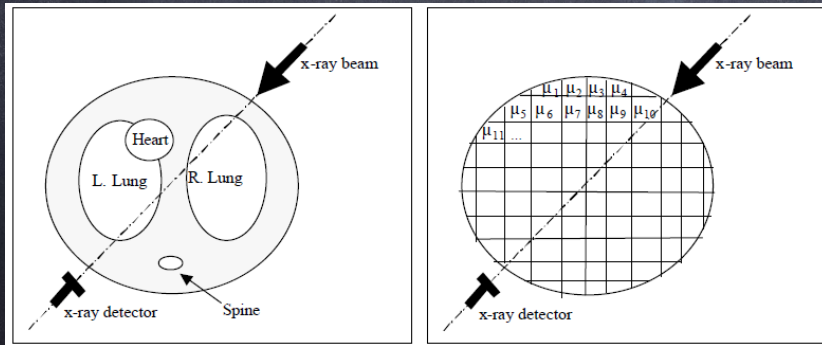
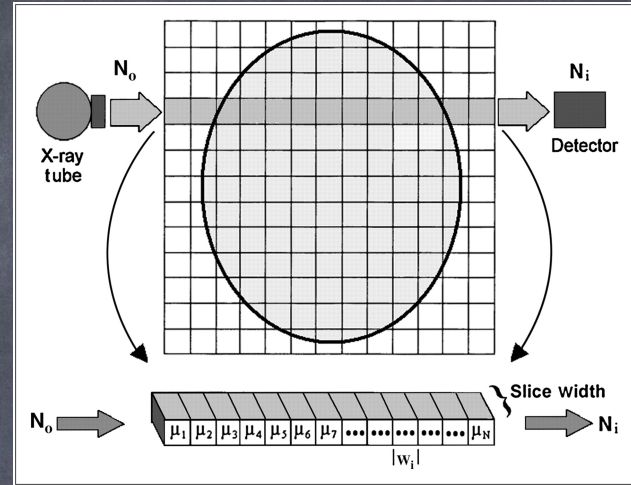
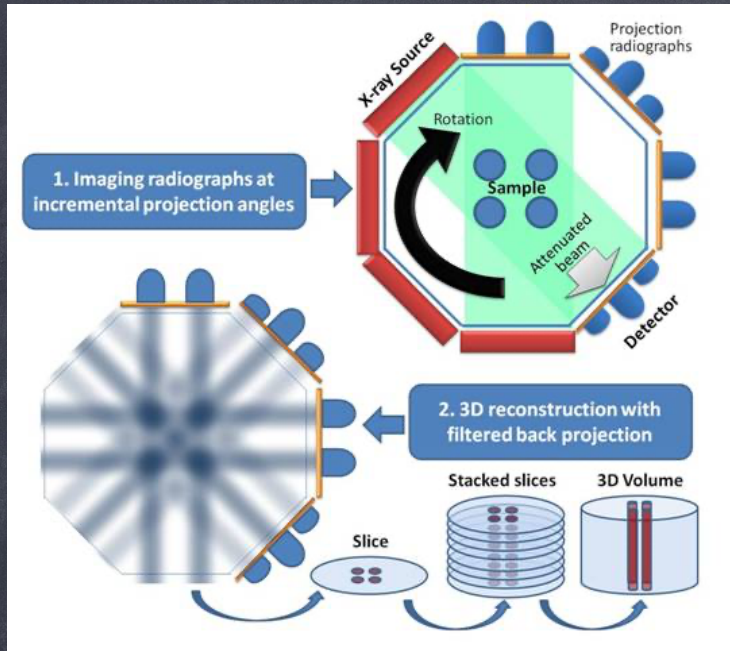
If we have N^2 pixels, ($N \times N$ grid of pixels)

Here $7 \times 7 = 49$

then we have N^2 unknowns. then we need N^2 equations to solve for the unknowns.

2D → 3D

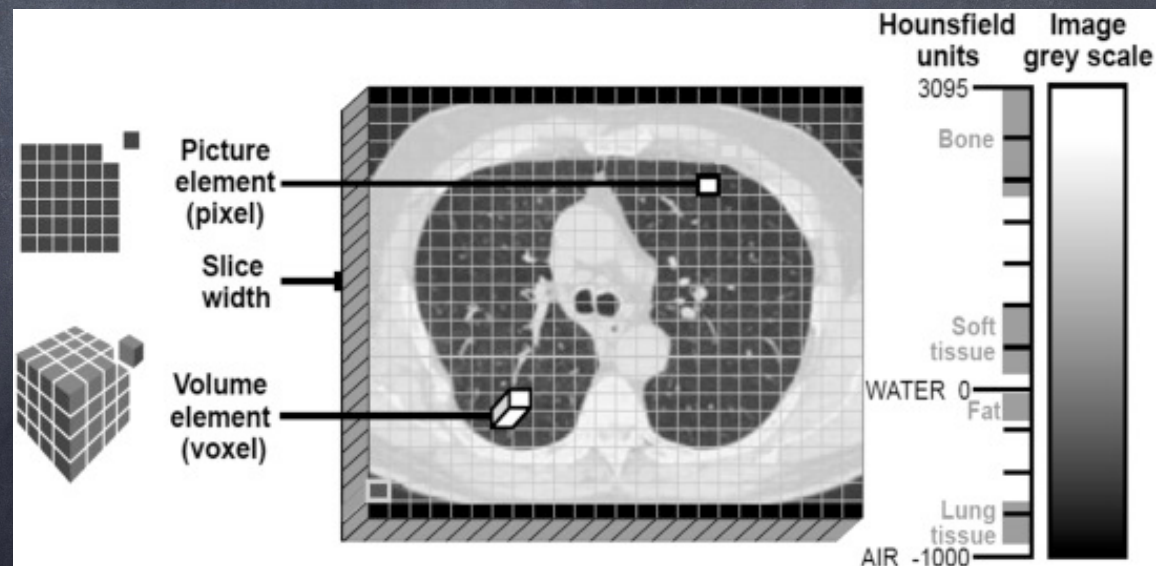




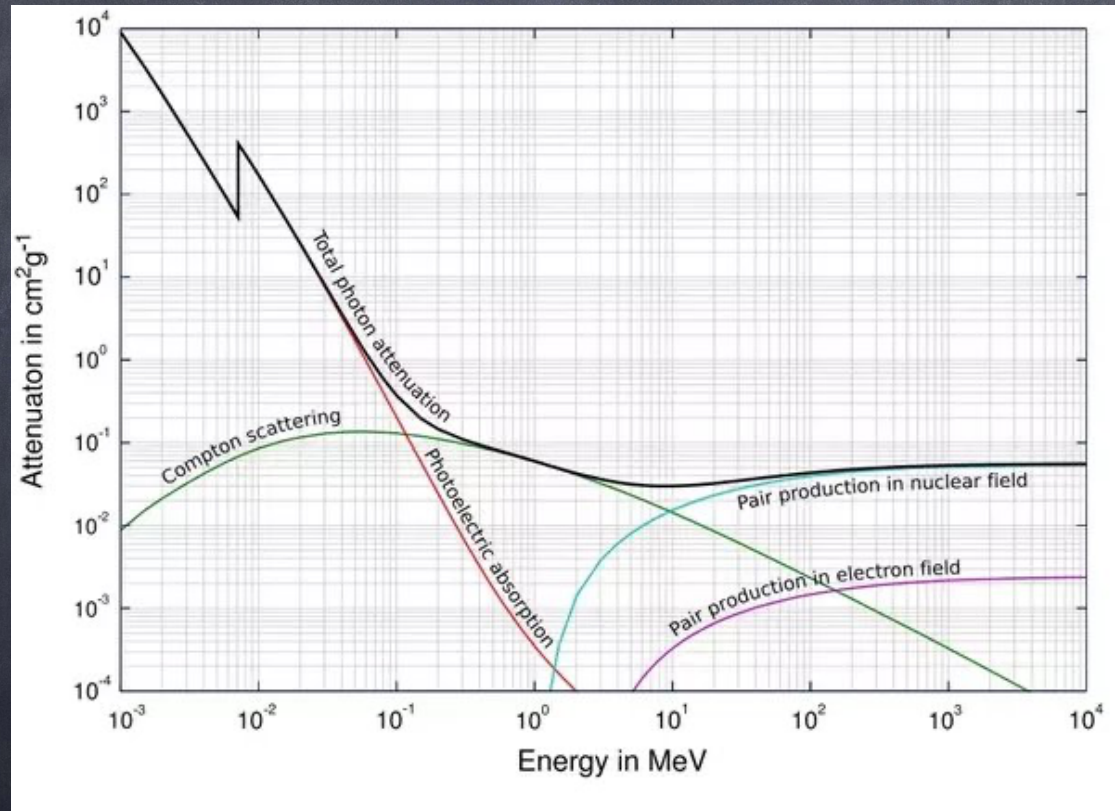
Typically, $N: 256 - 1024$

N^2 (# pixels) \sim 1 million unknowns

Typically, solved by computers.



Reminder of attenuation of γ -rays + gamma rays



x-rays

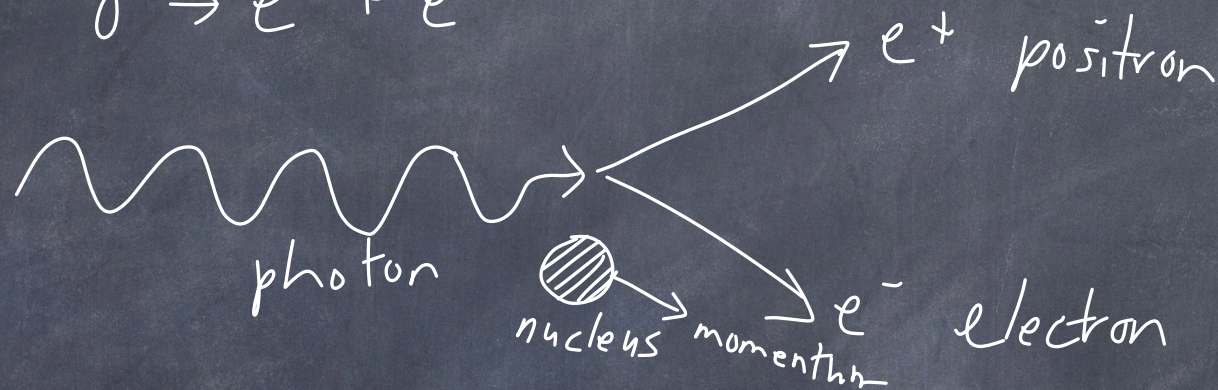


gamma rays

pair production starts here

If a photon has enough energy, it can convert its energy entirely to charged particles. The lightest charged particle is the electron.

$$\gamma \rightarrow e^- + e^+$$



This can't happen in free space, only near a massive object, such as a nucleus, such that the nucleus can supply momentum so that momentum is conserved.

How much ^{photon} energy is needed?

$$E = h\nu \geq 2m_e c^2 \quad m_e = 0.511 \frac{\text{MeV}}{c^2}$$

$$E > 1.022 \text{ MeV}$$

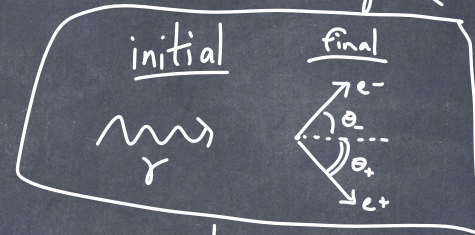
start S3: Supplementary proof that $\gamma \rightarrow e^+ + e^-$ can't happen in free space

we assume no nucleus!

initial: photon has energy: $h\nu$

final: no photon anymore,
electron has energy E_- and momentum \vec{p}_-
positron has energy E_+ and momentum \vec{p}_+

$m = m_e$: mass of electron



From conservation laws:

$$\text{Energy: } \underbrace{h\nu}_{\text{initial}} = \underbrace{E_+ + E_-}_{\text{final}} \quad (1)$$

$$\text{momentum } x: \frac{h\nu}{c} = p_- \cos \theta_- + p_+ \cos \theta_+ \quad (2)$$

$$\text{momentum } y: 0 = p_- \sin \theta_- + p_+ \sin \theta_+ \quad (3)$$

Rewrite (2) we get: $h\nu = cp_- \cos \theta_- + cp_+ \cos \theta_+$ (4)

Insert formula for relativistic energy ($E^2 = (cp)^2 + (mc^2)^2$) into (1):

$$h\nu = \sqrt{(cp_+)^2 + (mc^2)^2} + \sqrt{(cp_-)^2 + (mc^2)^2} \quad (5)$$

The maximum value of $h\nu$ in (4) is when $\cos\theta_- = \cos\theta_+ = 1$.

Then (4) becomes $h\nu = cp_- + cp_+$ (6)

But if we look at eq. (5), we see that

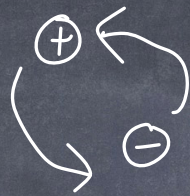
$(h\nu)^2$ must be greater than $(cp_-)^2 + (cp_+)^2$ because of the electron + positron masses.

Therefore, since we have 2 equations, (5) and (6), which can't be both true at the same time, this reaction is not valid, because energy + momentum can't be conserved simultaneously.

end

(53) finished

What happens to the positrons?



They orbit each other
(positron exists for
about 10^{-10} s)

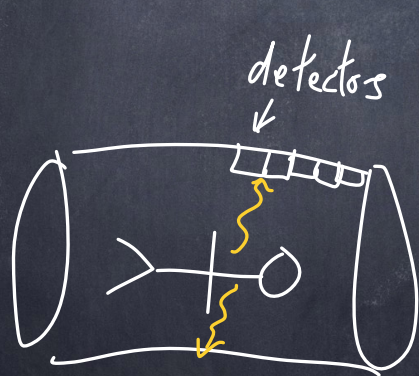
Then they annihilate:



(Two photons instead of one
because of momentum + energy
conservation)

This is the principle of positron emission tomography (PET).
we have to start with a source of positrons.

1) A positron-emitting radioactive element ^{15}O , ^{11}C , ^{13}F , ^{68}Ga is attached to a pharmaceutical + (ingested) (or injected)



Radioactive elements are usually prepared at an accelerator.

Two photons leave the body back to back. Detectors

that are 180° apart that look for the arrival of 2 coincident photons of 511 KeV.

By collecting data at different angles, we can make 3D images

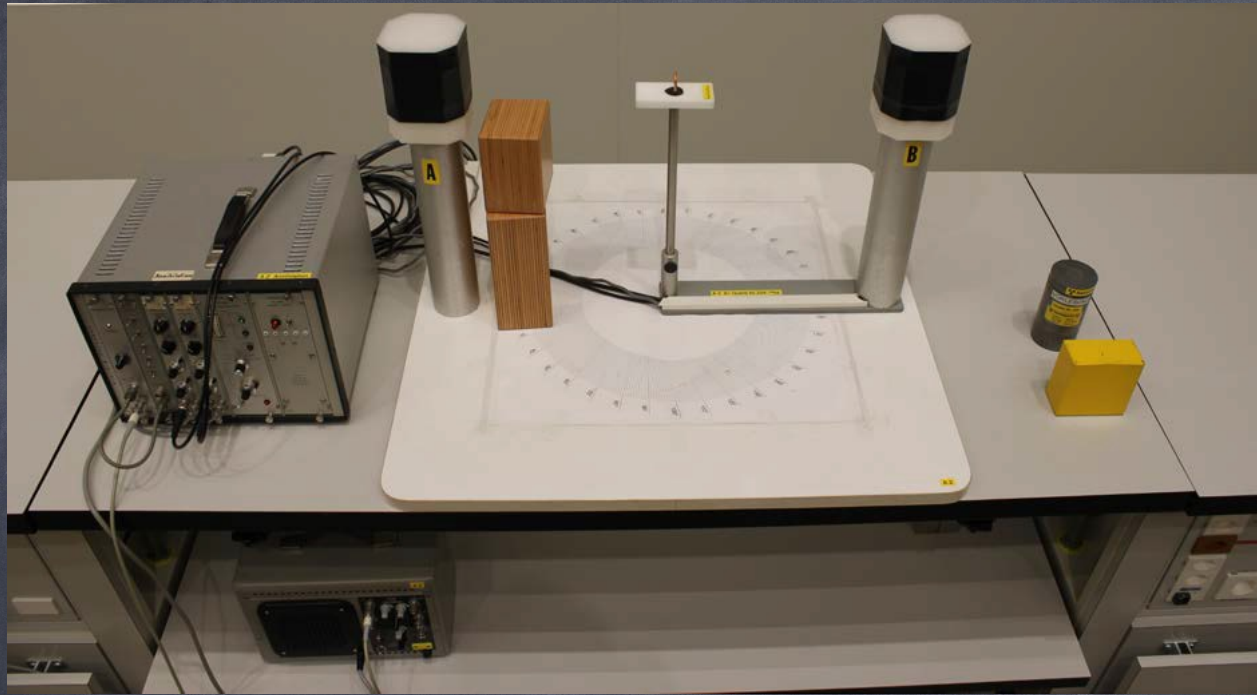
spatial resolution limited to about ~ 5 mm
because:

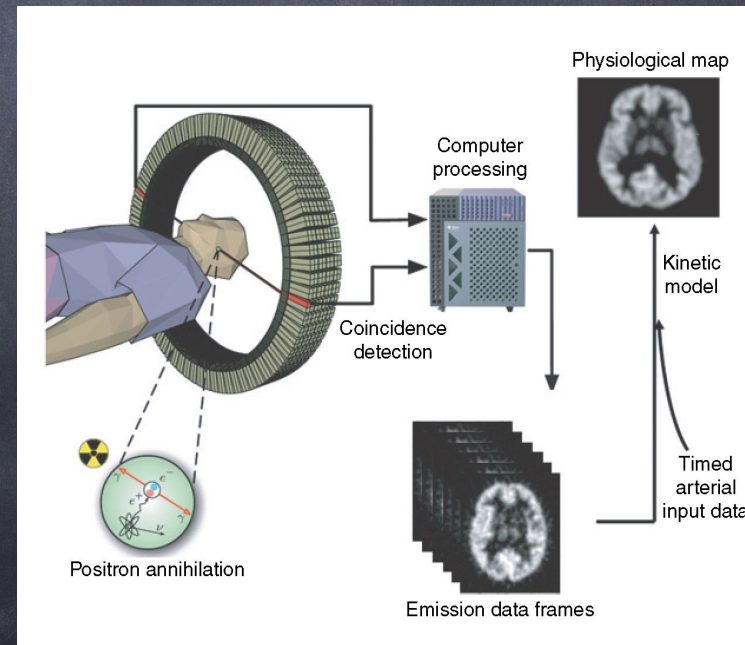
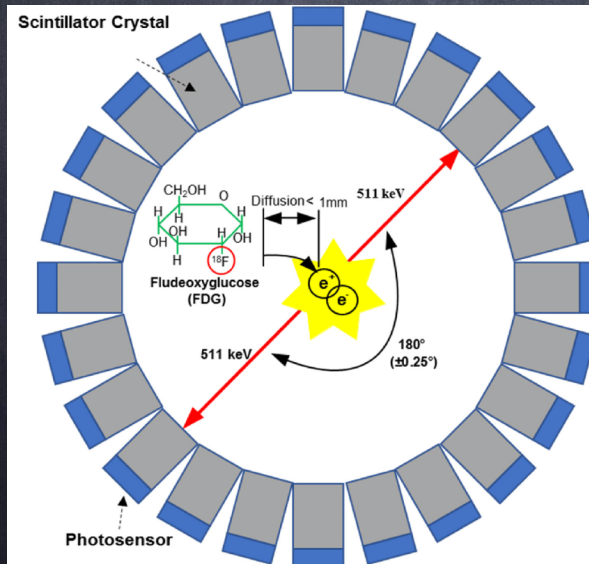
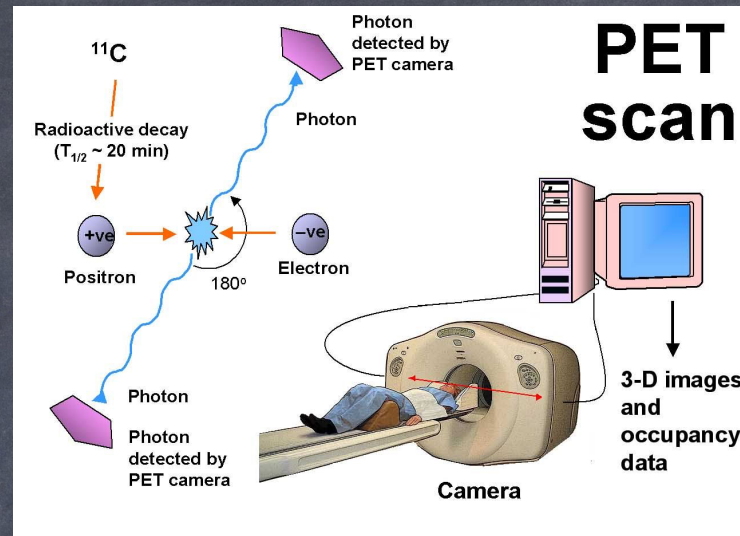
- 1) positronium (e^+e^-) has some non-zero momentum so the angle isn't exactly 180° .
- 2) positron can travel ~ 1 mm before it annihilates.

But PET scans can be done in real time.

By correlating images of blood flow or glucose or oxygen metabolism + monitoring a patient in some way, biochemical events can be correlated with brain activity.
(can reveal abnormal brain function)

PET is best used for monitoring time-dependent of metabolism of radiopharmaceuticals,
but not best for spatial resolution.





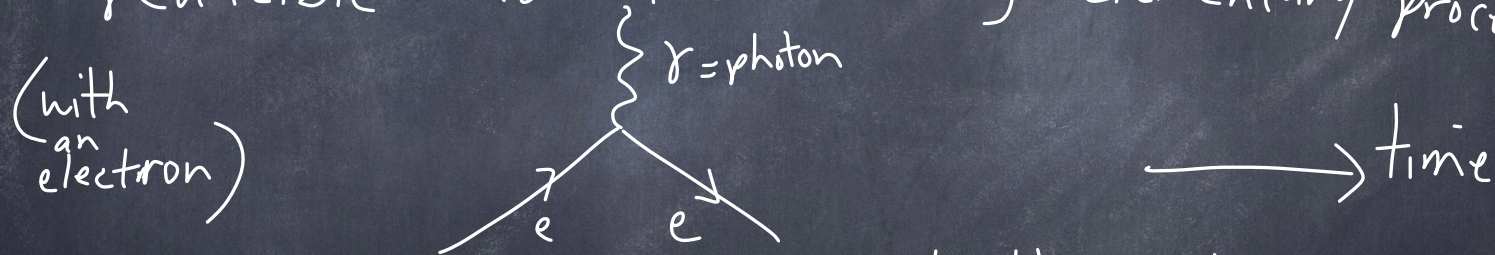
Supplement on Feynman
diagrams follows

Light is an electromagnetic wave.

Light is quantized. The unit of light is a photon.

Quantum electrodynamics (QED)

All electromagnetic phenomena are ultimately reducible to the following elementary process.



Time flowing horizontally to the right
This diagram reads "an electron enters, emits or absorbs a photon, and exits."

This diagram can be flipped or rotated,
and the process still happens.



A particle moving backwards in time is interpreted
as an antiparticle moving forwards in time.

electron = e^-

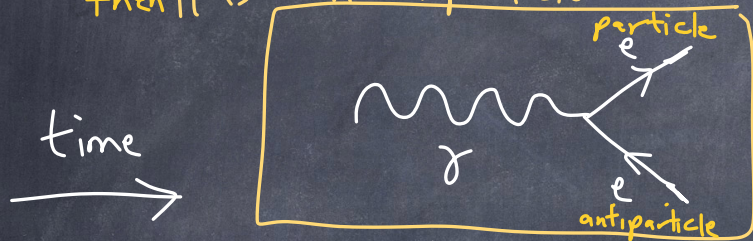
positron = e^+ antiparticle of the electron

A photon does not need an arrow since it is
its own antiparticle.

So this diagram reads " a positron enters,
emits or absorbs a photon,
and exits. "

The positron was predicted in 1928 by Dirac because his formulas had 2 solutions: $+$, $-$

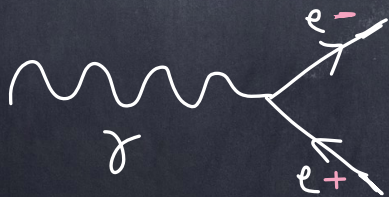
If arrow is moving opposite to time, then it is an antiparticle. Discovered in 1932 by Anderson



Can happen but must obey energy & momentum conservation.

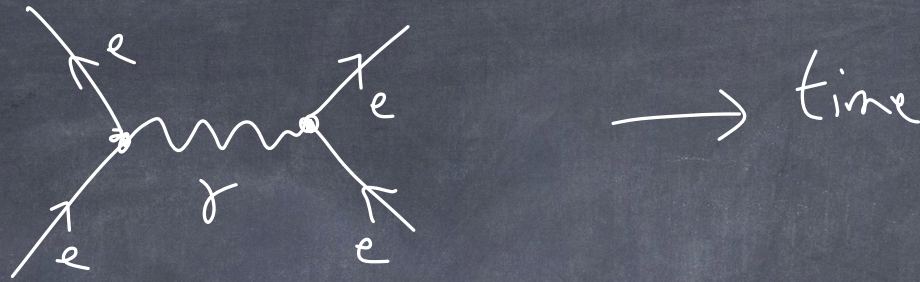
"A photon enters, decays into an electron and a positron, and they exit."

These diagrams are called Feynmann diagrams.



Some people label these diagrams with $e^- + e^+$, but I find this dangerous.

Since a positron moving backwards in time would be an electron



Here, an electron and positron annihilate into a photon, and then the photon decays into a new electron and positron.

Note: the electric charge is conserved.

We can write this diagram as:

$$e^- + e^+ \rightarrow \gamma \rightarrow e^- + e^+$$

electric charge $-1 + 1 = 0 = \overset{\uparrow}{0} = -1 + 1$

Energy + momentum are conserved in this process.

This diagram can be rotated.

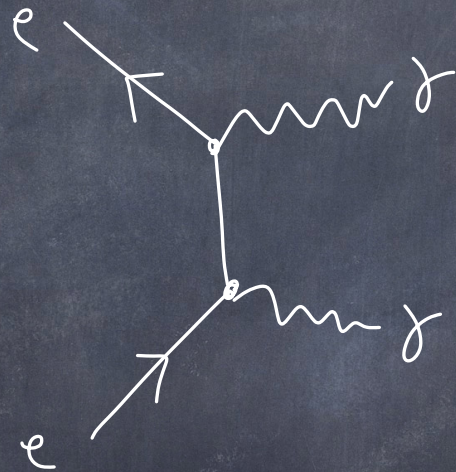


Here, two electrons enter, exchange a photon and continue as electrons.

Here, the electrons repel can be seen to repel each other.

In quantum physics, forces are mediated by particles. The photon mediates the electromagnetic force.

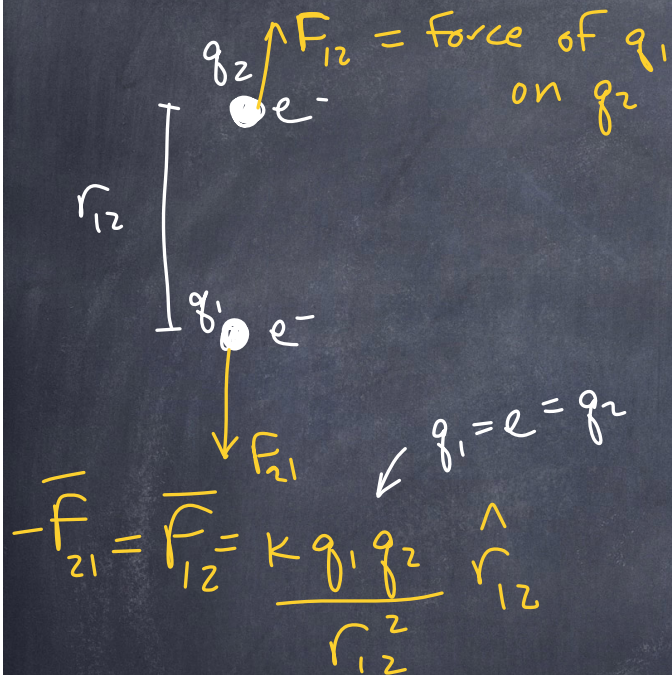
How do we draw the Feynman diagram for electron-positron annihilation (as in PET scan)?



read: an electron and positron annihilate into 2 photons.

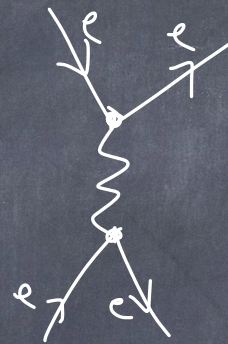
But does this mean classical physics is wrong?

classical physics view:



Electrons are repelled by a force, which we can calculate.

quantum physics view: $\xrightarrow{\text{time}}$



Here, two electrons exchange a photon and continue as electrons.

Here, the electrons repel each other.

In quantum physics, forces are mediated by particles. The photon mediates the electromagnetic force.

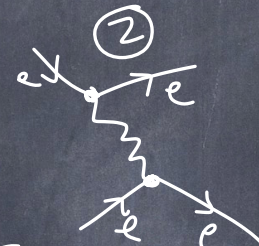
In practice, classical physics is easier to calculate for most everyday situations

What happens if a particle moves perpendicular to time?

A: Here a photon moves vertically.



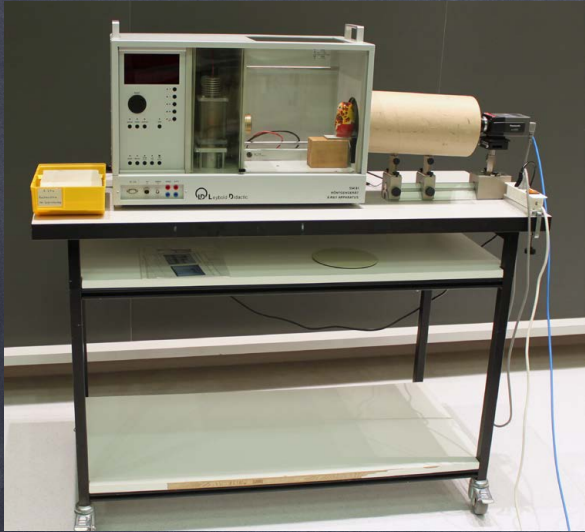
Really, what is happening is both of these:



In ①, the photon is emitted from the below electron.
In ②, the photon is emitted from the above electron.

The photon is not observable, we call it virtual.
we can't tell if ① or ② happens, so we use quantum mechanics to consider both.
But we draw it like this

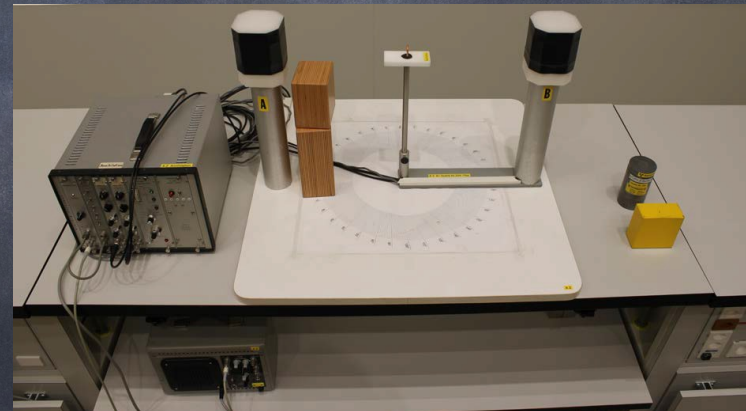




A24



ES90



A2